

# CO<sub>2</sub> Capture and Mineralisation

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Stephen B. Harrison, sbh4 GmbH

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# Renewable electricity can avoid CO<sub>2</sub> emissions from fossil fuel power generation and decarbonise many other sectors



Power and  
energy  
transition





- But not all sectors can be decarbonised with green electrons!
- Some industrial processes release **unavoidable** geogenic, biogenic or process CO<sub>2</sub> emissions that must be decarbonised using carbon capture.

# Geogenic CO<sub>2</sub> emissions from glass melting must be decarbonised



# Geogenic CO<sub>2</sub> emissions from the calcination of magnesite and dolomite for refractory materials must be decarbonised



Magnesite quarry for refractory materials production



# Geogenic CO<sub>2</sub> emissions from the calcination of limestone in lime and cement making must be decarbonised



# Process emissions from iron ore reduction in blast furnaces for iron making must be decarbonised





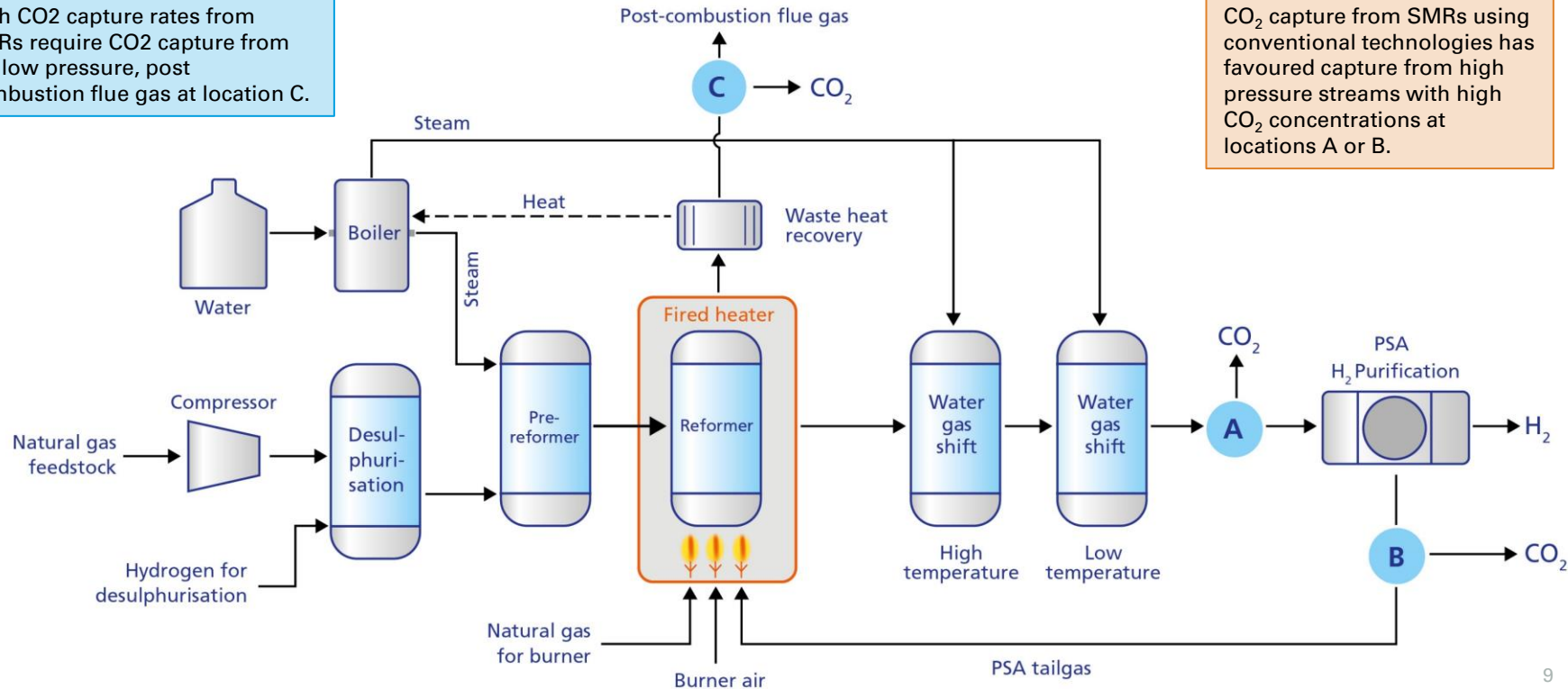
More than 1,000 SMRs produce grey hydrogen with process CO<sub>2</sub> emissions, many will be decarbonised





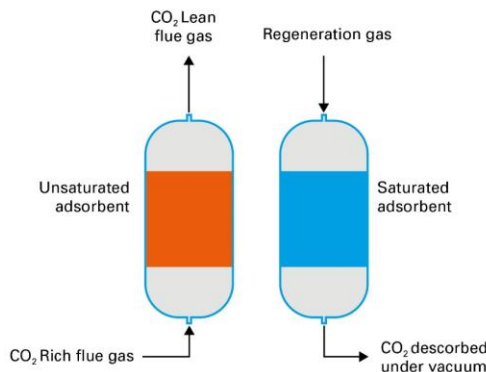
# Conventional CO<sub>2</sub> capture technologies benefit from high pressure, but for high capture rates, low pressure post-combustion flue gas must also be processed

High CO<sub>2</sub> capture rates from SMRs require CO<sub>2</sub> capture from the low pressure, post combustion flue gas at location C.

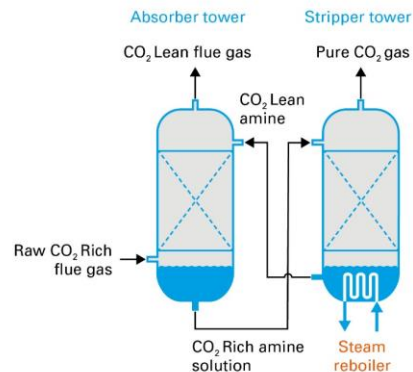


CO<sub>2</sub> capture from SMRs using conventional technologies has favoured capture from high pressure streams with high CO<sub>2</sub> concentrations at locations A or B.

# Conventional processes for CO<sub>2</sub> capture from SMRs and other industries require heat or power, in very large quantities



Change-over valves alternate the regeneration gas & the flue gas flow from one bed to the other.



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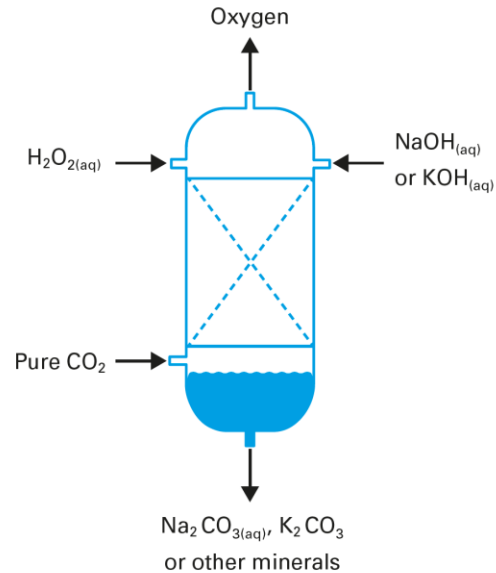
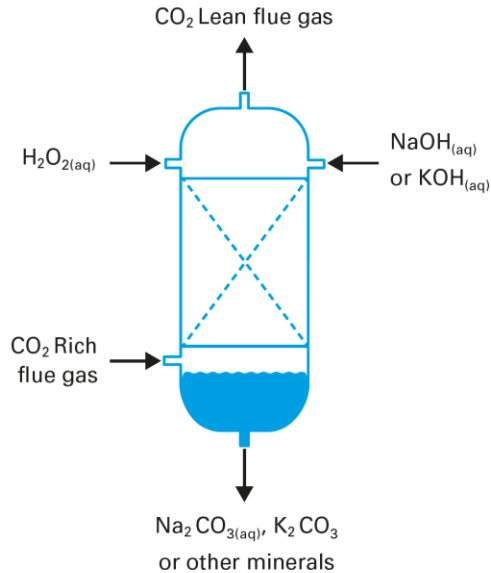
	VSA – vacuum swing adsorption	Amine Solvent with tower contactors
Separation principle	Adsorption	Absorption
Specific energy demand	1.7 GJ/t <sub>CO2</sub> (mostly power)	3 GJ/t <sub>CO2</sub> (mostly heat from steam)
Typical temperature	40°C	40-60°C in absorber, 120°C in stripper
Typical pressure	Cycling between moderate pressure and vacuum	Ambient to 30 bar
Typical CO <sub>2</sub> removal	< 90 %	> 90 %
Typical CO <sub>2</sub> purity	< 95 %	> 99 %
Typical plant size (tonnes per year CO <sub>2</sub> removal)	> 1,000 - 500,000	40,000 - 4,000,000
Technology maturity level	Commercial with some demonstrations, eg Air Products Port Arthur SMRs, USA	Commercial from many suppliers

# Mineralisation of CO<sub>2</sub> using alkalis and the superoxide radical – a new approach to capture CO<sub>2</sub> to form commercial minerals





# Airovation Technologies patented CO<sub>2</sub> capture and mineralisation process – CCM

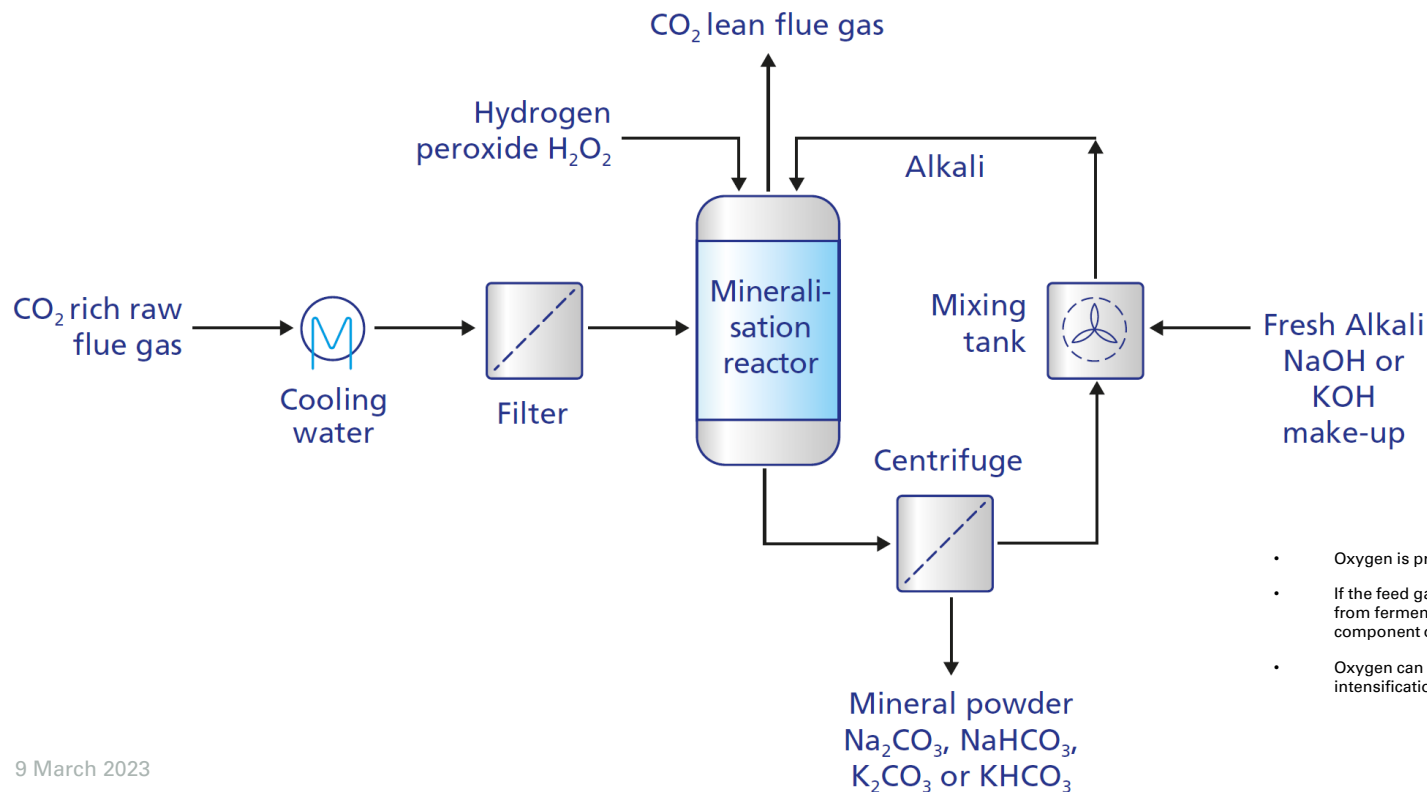


- CO<sub>2</sub> capture rate >95%
- Can capture CO<sub>2</sub> efficiently from low pressure, low concentration CO<sub>2</sub> streams in addition to higher pressure, higher concentration CO<sub>2</sub> flue gases
- Minimal heat and power requirement at the capture site
- Uses readily available commodity chemical feedstocks: NaOH or KOH, with H<sub>2</sub>O<sub>2</sub> being dosed as a minor component with catalytic properties
- Selectivity between commercial grade Na<sub>2</sub>CO<sub>3</sub>, K<sub>2</sub>CO<sub>3</sub>, NaHCO<sub>3</sub> or KHCO<sub>3</sub> as mineral products
- For very high concentration CO<sub>2</sub> streams (eg from fermentation and distillation), oxygen is produced (and can be utilised for process intensification in the bio-reactor)

# Mineralisation of CO<sub>2</sub> in lime water



# Airovation Technologies – patented CO<sub>2</sub> mineralisation process using alkalis and the superoxide radical



- Oxygen is produced during the process chemistry
- If the feed gas has a very high CO<sub>2</sub> concentration (eg from fermenters or bio-reactors), oxygen is the main component of the final flue gas
- Oxygen can be recovered and re-used for process intensification in bio-reactors and fermenters



# Airovation Technologies – alkali mineralisation of $\text{CO}_2$ using the superoxide radical



# Commercial carbonate and bicarbonate mineral products can be produced







# CCM – CO<sub>2</sub> mineralisation using alkalis and the superoxide radical to generate valuable materials

- Carbon capture and mineralisation (CCM) is an example of CCUS – carbon capture, utilisation and storage
- CO<sub>2</sub> emissions can be converted to minerals with commercial value
- The powders generated through mineralisation can be utilised commercially as chemical feedstocks
- The powders generated through mineralisation can be utilised as industrial feedstocks to conserve natural raw materials and reduce the energy intensity of mineral processing, eg cement and refractory materials
- Mineralisation using alkali solutions can avoid stressing power and heat utilities at the CO<sub>2</sub> capture site
- Moving feedstock and minerals to and from the CO<sub>2</sub> capture site uses standard logistics infrastructure
- The sweet spot for mineralisation using alkalis and the superoxide radical will exist in many situations...
  - CO<sub>2</sub> emission source of any pressure and any CO<sub>2</sub> concentration\*
  - Alkali and H<sub>2</sub>O<sub>2</sub> feedstocks are available
  - A market for the mineralisation product exists
  - A market for the oxygen exists (if a very high purity CO<sub>2</sub> stream is used)

# Example applications with ideal conditions: biotechnology and specialty glassmaking



- CO<sub>2</sub> emissions avoidance
- Biogenic CO<sub>2</sub> capture and conversion to bio- materials
- NaHCO<sub>3</sub> production for water softening, other marketable minerals possible
- O<sub>2</sub> production for bio-reactor process intensification



- CO<sub>2</sub> emissions avoidance
- Geogenic (unavoidable) CO<sub>2</sub> emissions capture and conversion to materials
- Na<sub>2</sub>CO<sub>3</sub> production as a feedstock for glassmaking (alternatively, other marketable minerals)
- Circularity in glassmaking materials



# Introduction to Stephen B. Harrison and sbh4 consulting

**Stephen B. Harrison** is the founder and managing director at sbh4 GmbH in Germany. His work focuses on decarbonisation and greenhouse gas emissions control. Hydrogen and CCUS are fundamental pillars of his consulting practice. He has served as the international hydrogen & CCS expert and team leader for multiple ADB projects related to renewable hydrogen deployment and CCS in several Asian nations.

With a background in industrial and specialty gases, including 27 years at BOC Gases, The BOC Group and Linde Gas, Stephen has intimate knowledge of hydrogen and carbon dioxide from commercial, technical, operational and safety perspectives. For 14 years, he was a global business leader in these FTSE100 and DAX30 companies.

Stephen has extensive buy-side and sell-side M&A due diligence and investment advisory experience in the energy and clean-tech sectors. Private Equity firms, investment fund managers and green-tech startups are regular clients.

As a member of the H2 View and **gasworld** editorial advisory boards, Stephen advises the direction for these international publications that focus on decarbonisation and CO<sub>2</sub> applications.

