



**TWO POWERHOUSES** HAVE JOINED FORCES.



## Oxygen enriched burners, CO<sub>2</sub> utilisation and mineralisation

By Stephen B. Harrison | 14 July 2021

**The global cement making industry is one of the largest emitters of carbon dioxide (CO<sub>2</sub>) and is responsible for four times more than the aviation sector.**

Annual global cement production in 2019 was 4.2 gigatonnes, of which 55% was produced in China. The world is projected to reach net zero by 2050, but the cement industry is expected to grow by 25% until 2050.

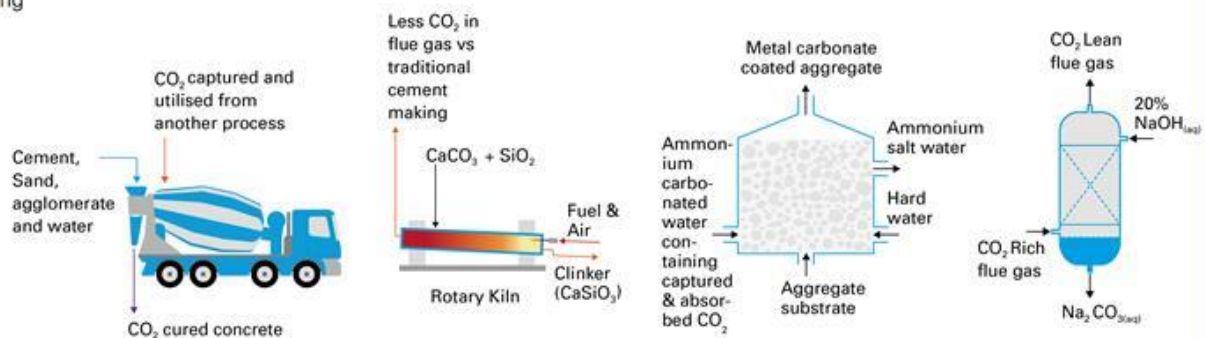
It is regarded as a 'hard-to-abate sector', as much of its CO<sub>2</sub> emissions are intrinsically connected to the chemical process of cement making. While electrification and the use of hydrogen can be mitigated using electrification with renewable power. At 900°C limestone is decomposed into lime and CO<sub>2</sub>.

In addition to CO<sub>2</sub> emissions from the process, there are also emissions associated with the burner in the cement kiln. Coal and petcoke are burned to make the heat and high temperature that are required to drive the chemical reactions.

Accounting for the mixing of CO<sub>2</sub> released from the mineral processing and combustion, the typical CO<sub>2</sub> concentration in the atmosphere is 415 ppm, which is 100 times higher than the CO<sub>2</sub> partial pressure in the atmosphere, it is still challenging to capture CO<sub>2</sub> from a flue gas stream.

### Mineralisation processes for CO<sub>2</sub> emissions reduction in cement making and concrete production

sbh4  
consulting



© 2021 sbh4 GmbH

	CO <sub>2</sub> cured concrete	New cement type based on calcium silicate instead of tricalcium silicate	Agglomerate recycling	Sodium bicarbonate formation
Mitigation principle	Cement use reduction through increased concrete strength and CO <sub>2</sub> utilisation	Reduced CO <sub>2</sub> flue gas emissions from cement clinker kiln and CO <sub>2</sub> utilisation	CO <sub>2</sub> mineralisation reaction with recycled or waste cements	CO <sub>2</sub> mineralisation from cement kiln and / or calciner flue gas
CO <sub>2</sub> Mitigation potential	CO <sub>2</sub> utilisation during concrete curing ~5% reduction from cement use reduction	CO <sub>2</sub> utilisation during concrete curing ~50% reduction from different cement chemistry	~90% reduction from CO <sub>2</sub> capture & mineralisation	~90% reduction from CO <sub>2</sub> capture
Carbonate formed	CaCO <sub>3</sub>	CaCO <sub>3</sub>	CaCO <sub>3</sub> and MgCO <sub>3</sub>	Na <sub>2</sub> CO <sub>3</sub>
Raw material	Ca(OH) <sub>2</sub> from fresh concrete	Ca(OH) <sub>2</sub> from fresh concrete	Ca(OH) <sub>2</sub> in re-used concrete, CaO and MgO from fly ash and steel slag	NaOH produced from electrolysis of NaCl
Technology maturity level	Commercial, eg CarbonCure	Commercial, eg Solidia	Demonstration, eg Blue Planet	Demonstration, eg SkyMine

the costs of the oxygen supply and equipment modifications.

An increased CO<sub>2</sub> concentration in the flue gases eventually makes carbon capture much more cost effective emissions such as NO<sub>x</sub>, SO<sub>x</sub> and particulates can be simplified using combustion with oxygen enriched air be downsized.

### **Separation of CO<sub>2</sub> streams from process and combustion off-gases**

An alternative approach is to use air to provide oxygen for the combustion process and separate the process Carbon capture can then be focused on the process gas stream which has a higher CO<sub>2</sub> concentration.

About 65% of the CO<sub>2</sub> emissions from cement making are associated with the process gas stream so there is decarbonisation.

Within the European project Low Emissions Intensity Lime and Cement (LEILAC) a pilot plant has been built . The 60m tall pilot plant has a capture capacity of about 18,000 tonnes of CO<sub>2</sub> per year which results from 24 production or 190 tonnes per day of ground limestone feedstock.

Core to the process is the direct separator reactor (DSR) which has been developed by Calix. It acts as a large thereby enabling separation of the combustion and process gas emissions streams. The exhaust of the process captured and liquefied for further utilisation.

A similar technology is also employed by CarbonEngineering within its Direct Air Capture (DAC) process.

The follow-up project LEILAC2 started in 2020. Intended to be operational by 2023 at the Hannover plant of 100,000 tonnes of CO<sub>2</sub> per year. If natural gas is employed as the fuel to heat the limestone, conventional can separate the CO<sub>2</sub> from the rest of the flue gas.

During the LEILAC2 project alternative fuels based on biomass and the use of electrical heating using renewable calciner. Furthermore, the flexible combination of intermittent renewable electricity and fuel is to be validated additional dispatchable power consumption can be offered as a service to the electricity grid.

One important question remains. Even if all anthropogenic CO<sub>2</sub> emissions from cement making plants are ca



formation. Some primary minerals like olivine, a weak basalt rock with the chemical formula  $(\text{Mg, Fe})_2\text{SiO}_4$ , <sup>4</sup> environmental conditions. Lessons from this natural weathering process can be used to capture CO<sub>2</sub> emission

To store CO<sub>2</sub> in a carbonate form, a raw material with a high metal oxide or metal hydroxide content is required which has a distinctive calcium hydroxide fraction. Other raw materials include fly ash (CaO 10-40%, MgO 0-10%). These materials also contain high concentrations of other metal oxides. For example, MgO can be utilised to form magnesium carbonates ( $\text{MgO} + \text{CO}_2 \rightarrow \text{MgCO}_3$ ).

### Growing underground rocks with CO<sub>2</sub>

To implement this mineralisation carbon capture process, two options are available. Either transport the raw materials to the CO<sub>2</sub> source or transport the CO<sub>2</sub> to suitable rock formations to store the gas securely.

The latter approach has been developed by CarbFix in Iceland since 2007. CO<sub>2</sub> captured from flue gases or dissolved in an aqueous solution of carbonic acid. This solution is pressurised and pumped underground where it heats in hot water. As hot water can bind less CO<sub>2</sub> than cold water, the CO<sub>2</sub> is released underground, and carbonation of basalt rocks takes place.

### Cement curing with CO<sub>2</sub>

The Canadian-based company Carbon Cure takes advantage of the sequestration capability of concrete by injecting CO<sub>2</sub> during concrete preparation. This increases the strength of the concrete as additional calcium carbonate is formed. The result is that 5% less cement is required to make the concrete, and a reduced carbon footprint for emissions from a reduced requirement for cement making.

The technology is readily available to reduce the CO<sub>2</sub> emissions of concrete and can be scaled up immediately in the production process. Several reference projects exist in the US, with the Amazon HQ2 the largest project to date.



To sequester anthropogenic CO<sub>2</sub> emissions, Blue Planet has developed a new mineralisation process. It uses concrete contains aggregate and an old cement fraction. During the process, the aggregate is upcycled and c Whereas the old cement fractions are mineralised with CO<sub>2</sub> to form a new layer of calcium carbonate around seeds for the mineralisation process.

The CO<sub>2</sub>-sequestered aggregate is used in addition to the upcycled aggregate in newly mixed concrete. In a used at the Interim Boarding Area B at San Francisco International Airport.



For  
The  
pro  
sec  
kiln  
wh  
hyc  
hyc  
pro  
knc  
  
In a  
of t  
pla  
ope  
CO  
pro  
Ult  
atn

About the author

Stephen B. Harrison is Managing Director of sbh4 consulting. Harrison has over 30 years' experience of the